

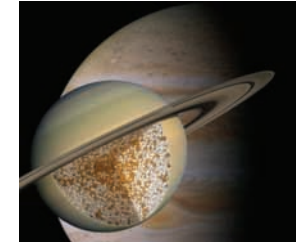
First Principles Simulations of Hydrogen and Helium at High Pressure

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Outline



- Introduction and Motivation
- Simulation Methods
- CEIMC Simulations
 - Metallic hydrogen EOS
 - DFT benchmark
- Hydrogen-Helium Mixtures
 - Equation of state
 - Phase separation
- Molecular Dissociation
 - 1st order liquid-liquid phase transition
 - H₂ melting line

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First-Principles Simulations: direct solution of electronic problem

- Fix the position of the nuclei, solve the Schrödinger eqn. for the electrons (BO approximation):

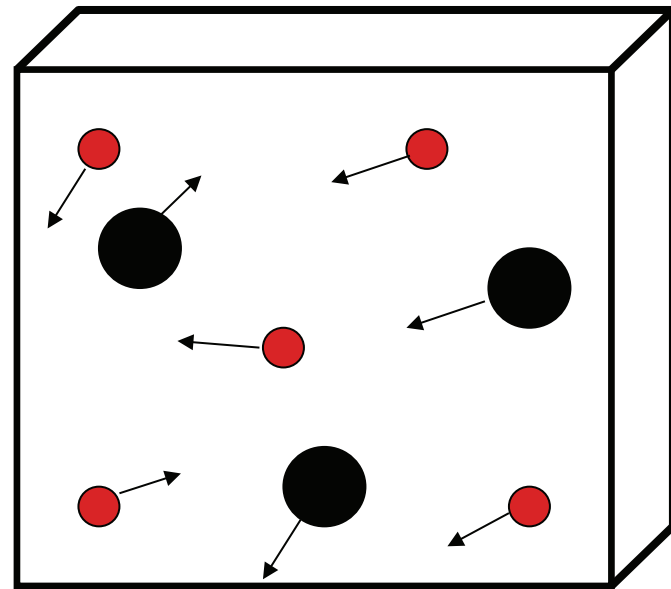
$$\left[-\frac{\hbar^2}{2m} \vec{\nabla}^2 + V(\vec{r}, \{\vec{R}\}) - E_{BO} \right] \Psi(\vec{r}) = 0$$

$$\hat{V}(\vec{r}, \{\vec{R}\}) = \hat{V}_{e-e} + \hat{V}_{e-n} + \hat{V}_{n-n}$$

- Electronic Structure Methods:

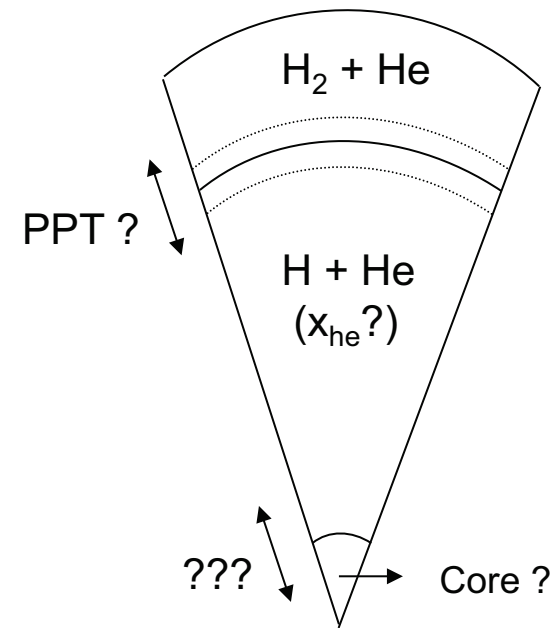
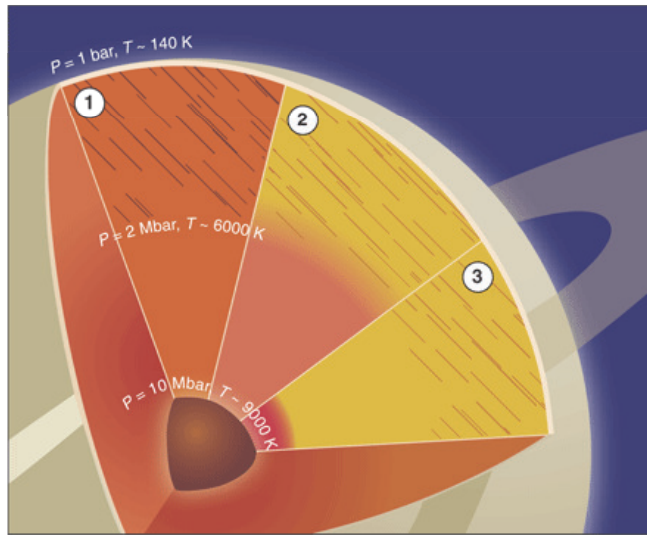
- Tight-binding, non-SCF DFT
- Density Functional Theory (DFT)
 - Semi-local functionals
 - Hybrids
- Quantum Monte Carlo (QMC)
- Quantum Chemistry

Accuracy
↓



- Nuclei represent an external potential in the electronic problem

Motivation



■ Saturn's Luminosity

- Homogeneous evolutionary models do not work for Saturn
 - Predicted age: $\sim 2\text{-}2.7 \text{ Gyr}$
- Additional energy source in planet's interior is needed
 - Helium segregation
 - Gravitational differentiation

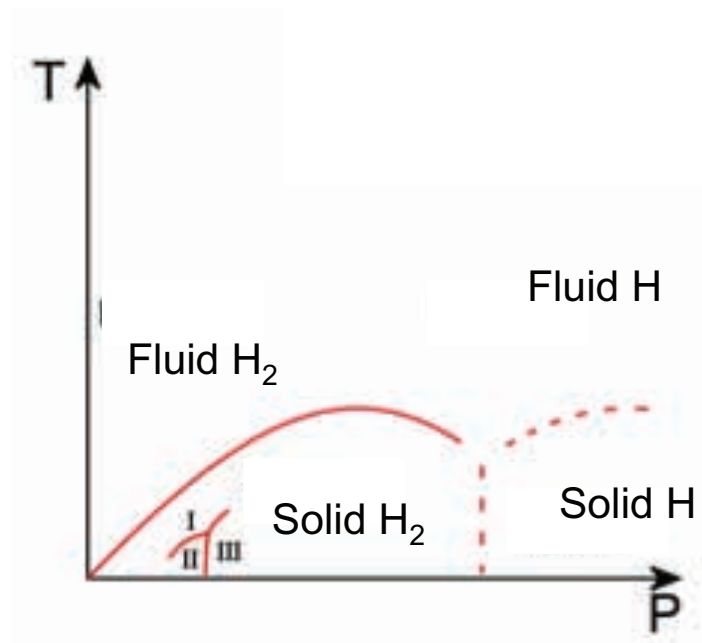
■ Giant Planets

- Primary components
- $P(\rho, T, x_i)$ closes set of hydrostatic equations
- Interior models depend very strongly on EOS and phase diagram

Figure taken from: Fortney J. J., *Science* **305**, 1414 (2004).

Hydrogen Phase Diagram

- Interesting physics:
 - Molecular dissociation
 - Metal-Insulator transition
 - Melting



CEIMC simulations

Free Energy Calculations

- Free energies are hard to calculate!!!
 - Not ensemble averages, they are related to the available phase space
- Direct evaluation using first-principles is too expensive
 1. Build simple effective potential that resembles DFT / QMC.
 2. Do complicated calculations on the simple system
 3. Calculate free energy difference using Coupling Constant Integration
 4. Expand free energy using EOS from regular simulation

$$\frac{F(V, T_2, x)}{T_2} - \frac{F(V, T_1, x)}{T_1} = - \int_{T_1}^{T_2} \left(\frac{E(V, T, x)}{T} \right) \frac{dT}{T}$$

$$F(V_2, T, x) - F(V_1, T, x) = - \int_{V_1}^{V_2} P(V, T, x) dV$$

$$V(\lambda) = \lambda V_1 + (1 - \lambda) V_0$$

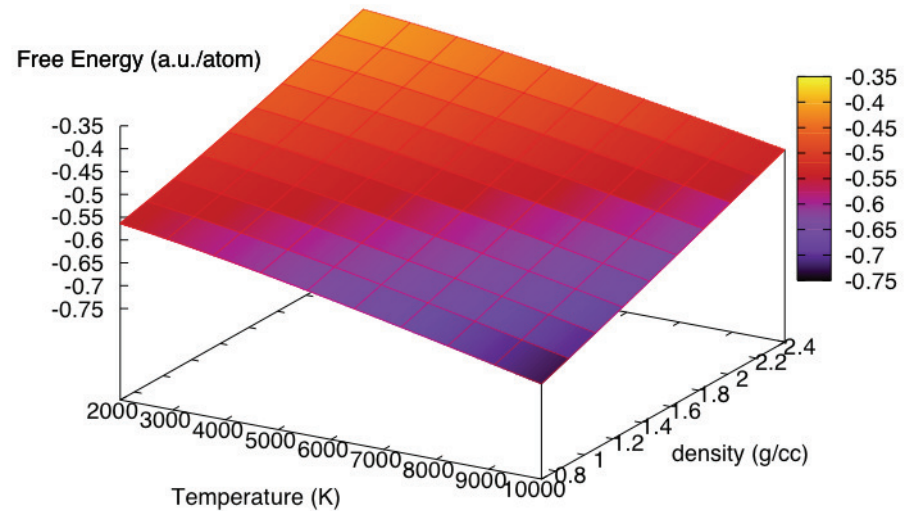
$$\begin{aligned} F_1(T, V, N) - F_0(T, V, N) &= \int_0^1 d\lambda \left(\frac{dF(\lambda)}{d\lambda} \right) \\ &= \int_0^1 d\lambda \langle (V_1 - V_0) \rangle_{T, V, N, \lambda} \end{aligned}$$

Very efficient method for simple systems

Free Energy Fits

- Free energy integration with CEIMC
- χ^2 fit to QMC results:

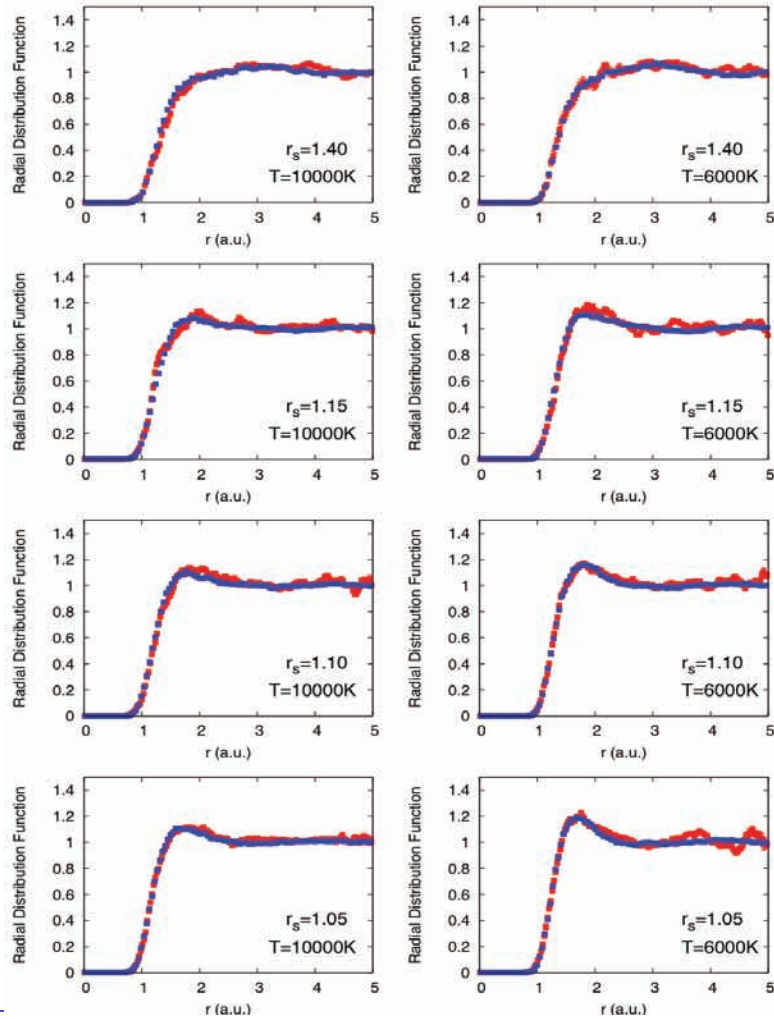
$$F(T, \rho) = \sum_{i,k} c_{i,k} g_i(T) g_k(\rho)$$



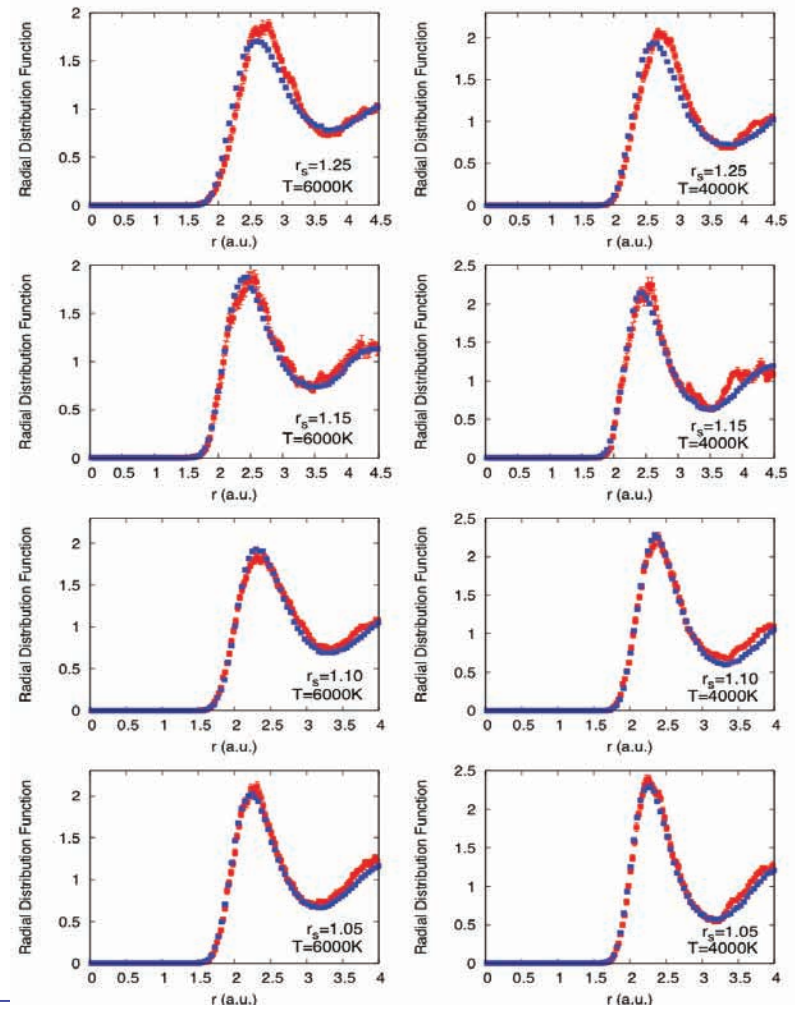
Details in: Phys. Rev. E **81**, 021202 (2010)

Radial Distribution Functions Agree

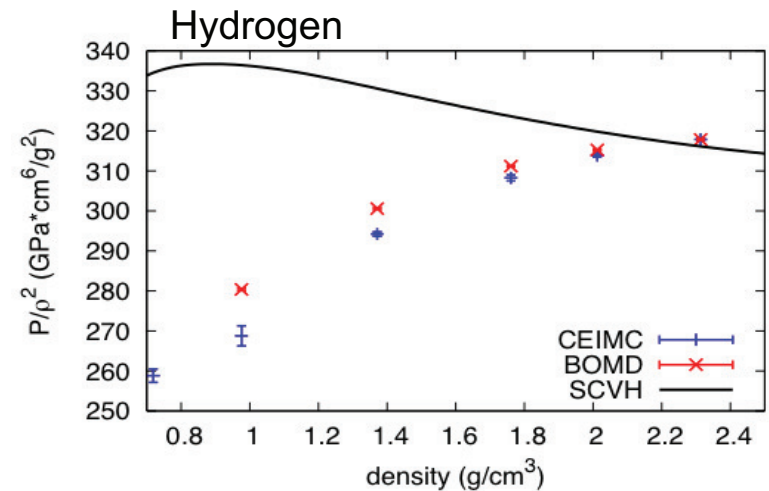
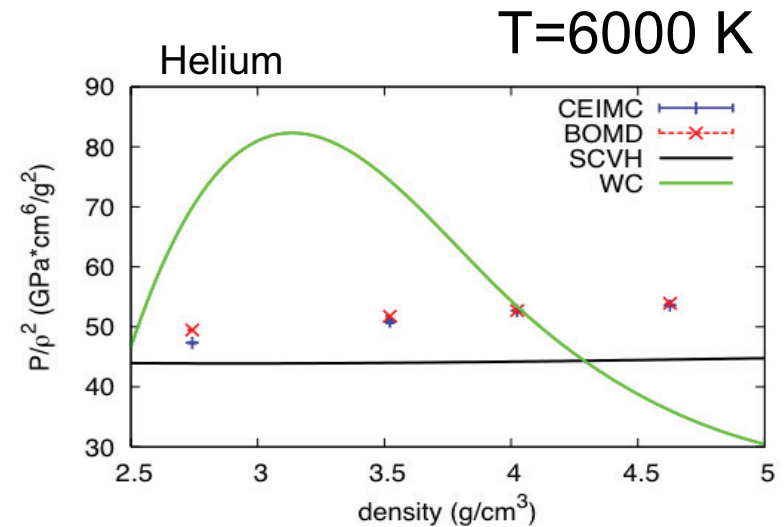
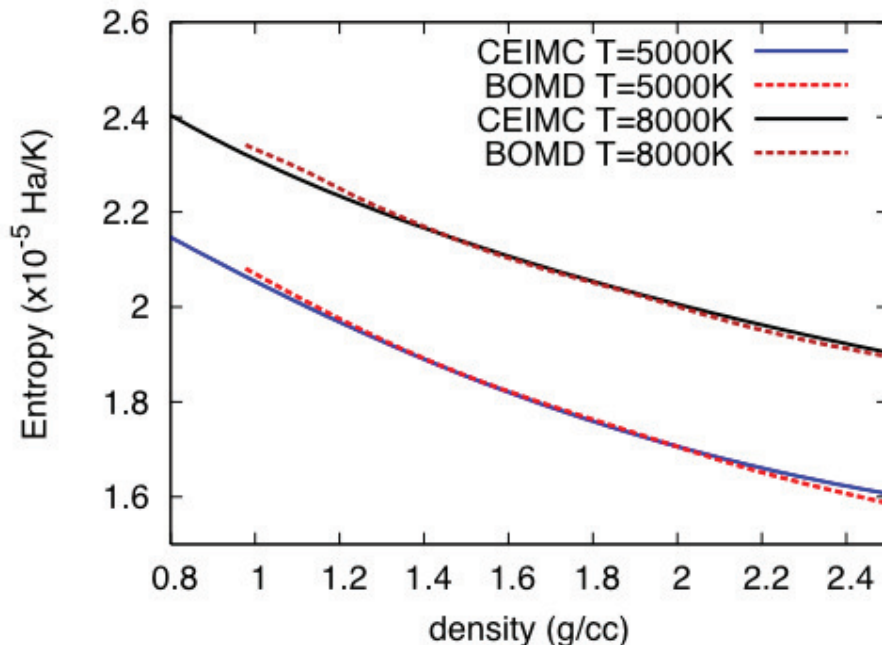
Hydrogen



Helium



EOS Comparison



- Excellent agreement with DFT
- Poor agreement with chemical models

H-He mixtures

Previous work

Fully ionized models

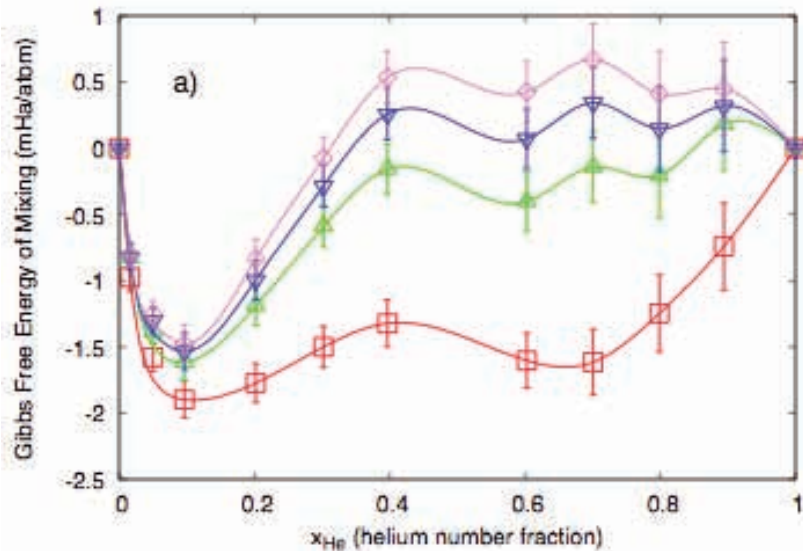
- Stevenson 1975, Hubbard-DeWitt 1985, Pollock-Alder 1976, etc.
 - Protons + Alpha particles in a uniform compensating negative background
 - Low demixing temperatures → no phase separation in planets
 - Predict $T_m(P)$ with negative slope

First Principles

- Ideal mixing approximation
 - Klepeis, *et al.* - 1990: $T_m \sim 15,000$ K → major differentiation
 - Mixing Enthalpy from calculations on alloys of H-HE
 - Pfaffenzeller, *et al.* - 1994: $T_m \sim 4000 - 6000$ K → no phase separation
 - Improved over Klepeis, *et al.* by allowing structural relaxation
 - Redmer, *et al.* - 2009: $T_m \sim 8000 - 9000$ K
 - Composition dependence of enthalpy by BOMD.

Mixing Free Energy

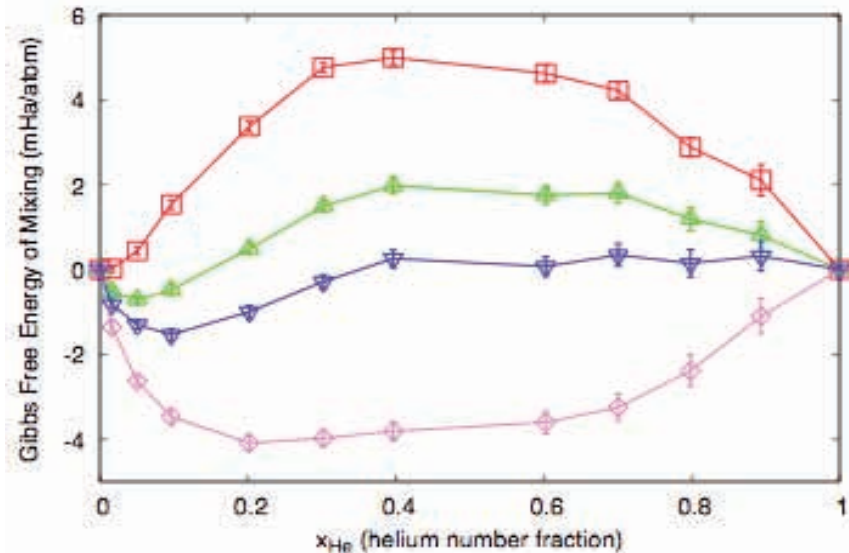
T=8000 K



— 4 Mbar — 8 Mbar
— 10 Mbar — 12 Mbar

- Clear minimum at low helium fraction.

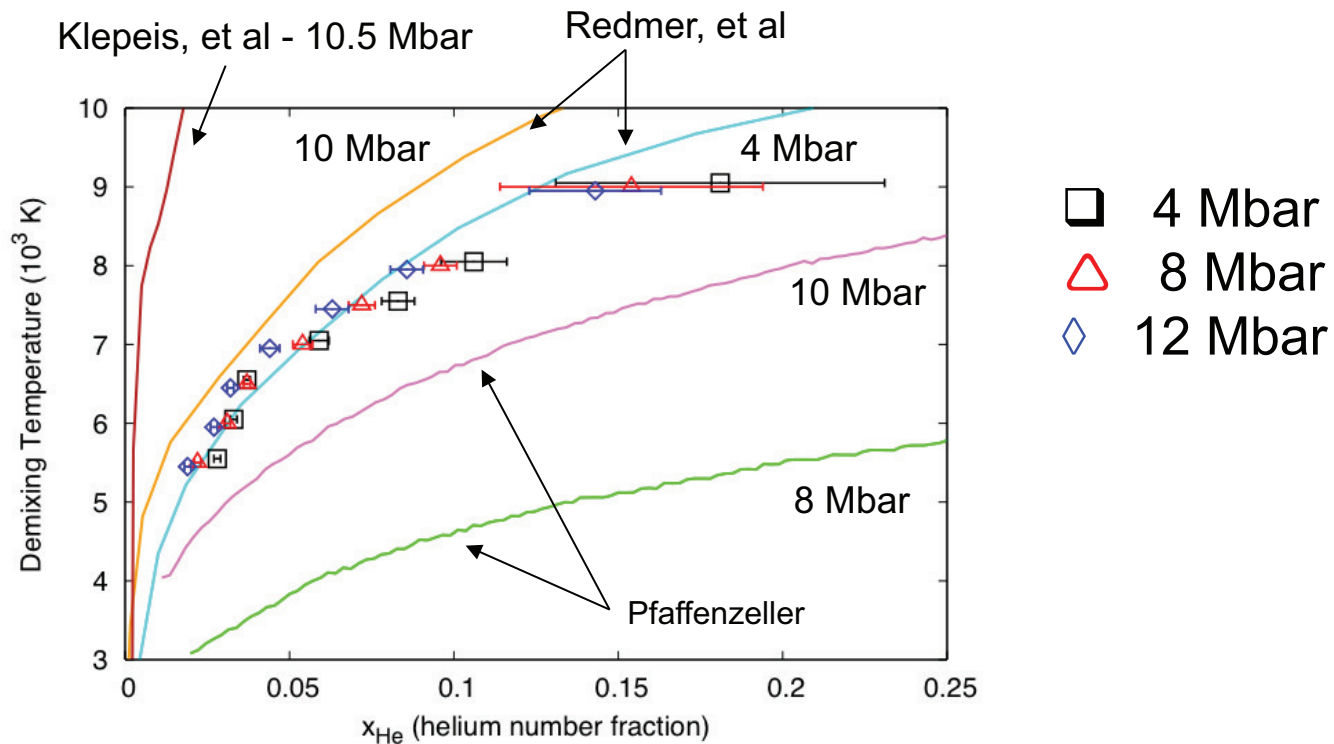
P=10 Mbar



— 4000 K — 7000 K
— 9000 K — 10000 K

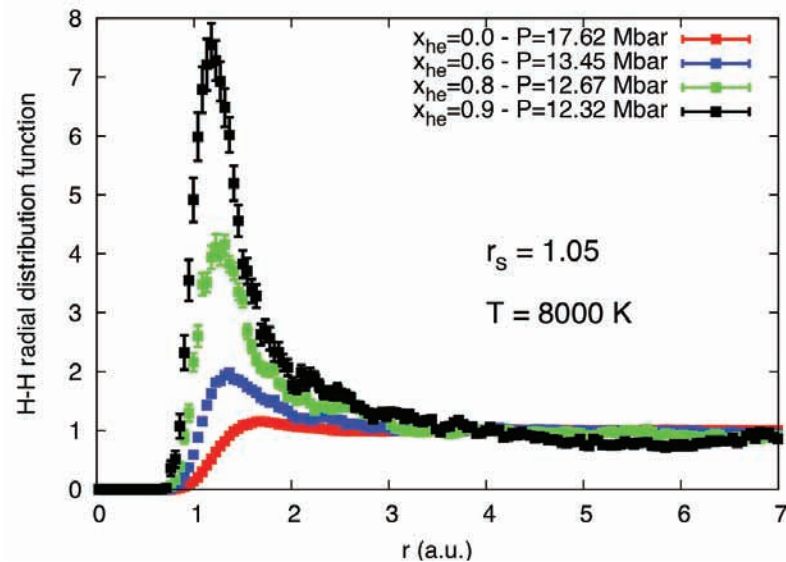
- Very strong temperature dependence, fairly insensitive to pressure.

Demixing Temperature

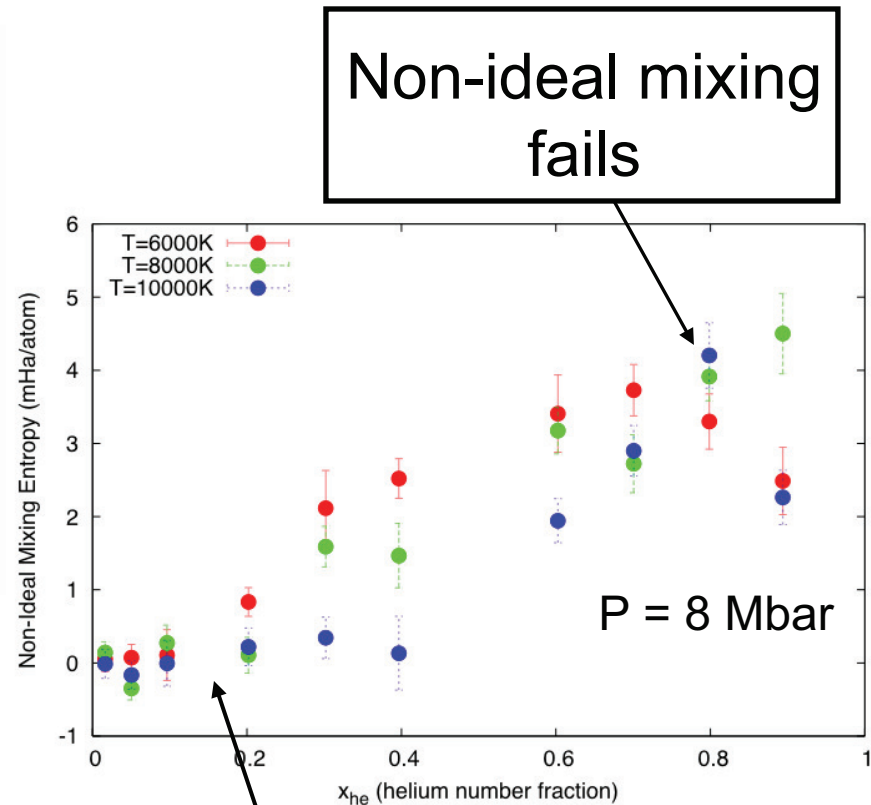


- Previous CPMD simulations underestimate demixing temperature.
- Good agreement with Redmer *et al.*
 - Differences come from non-ideal effects

Molecular-like Correlations

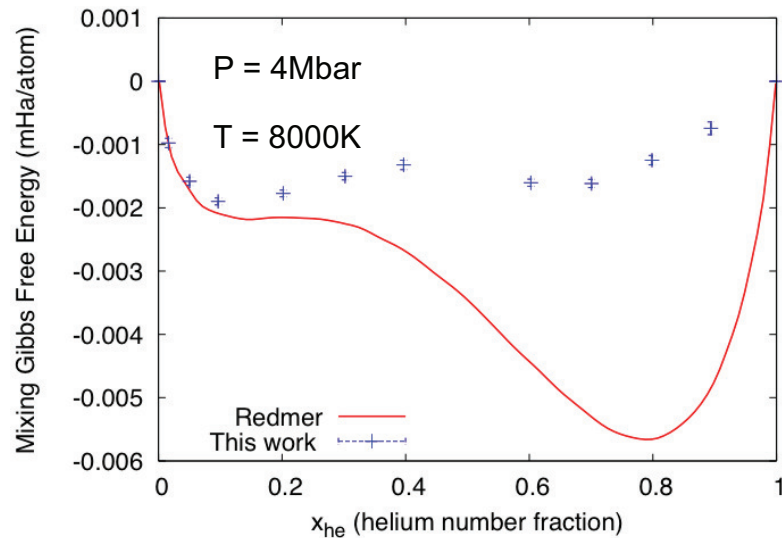


- Weak attraction even at very high pressures
- Induces molecular-like correlations
 - Pseudo-molecular state has smaller entropy compared to atomic state.

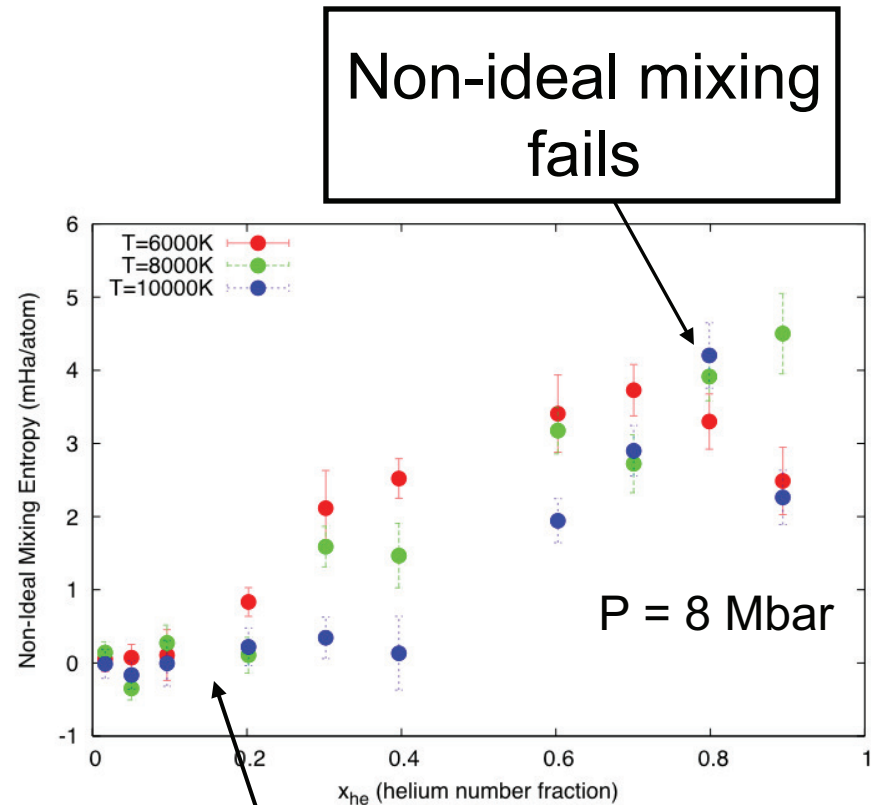


Ideal mixing works

Molecular-like Correlations

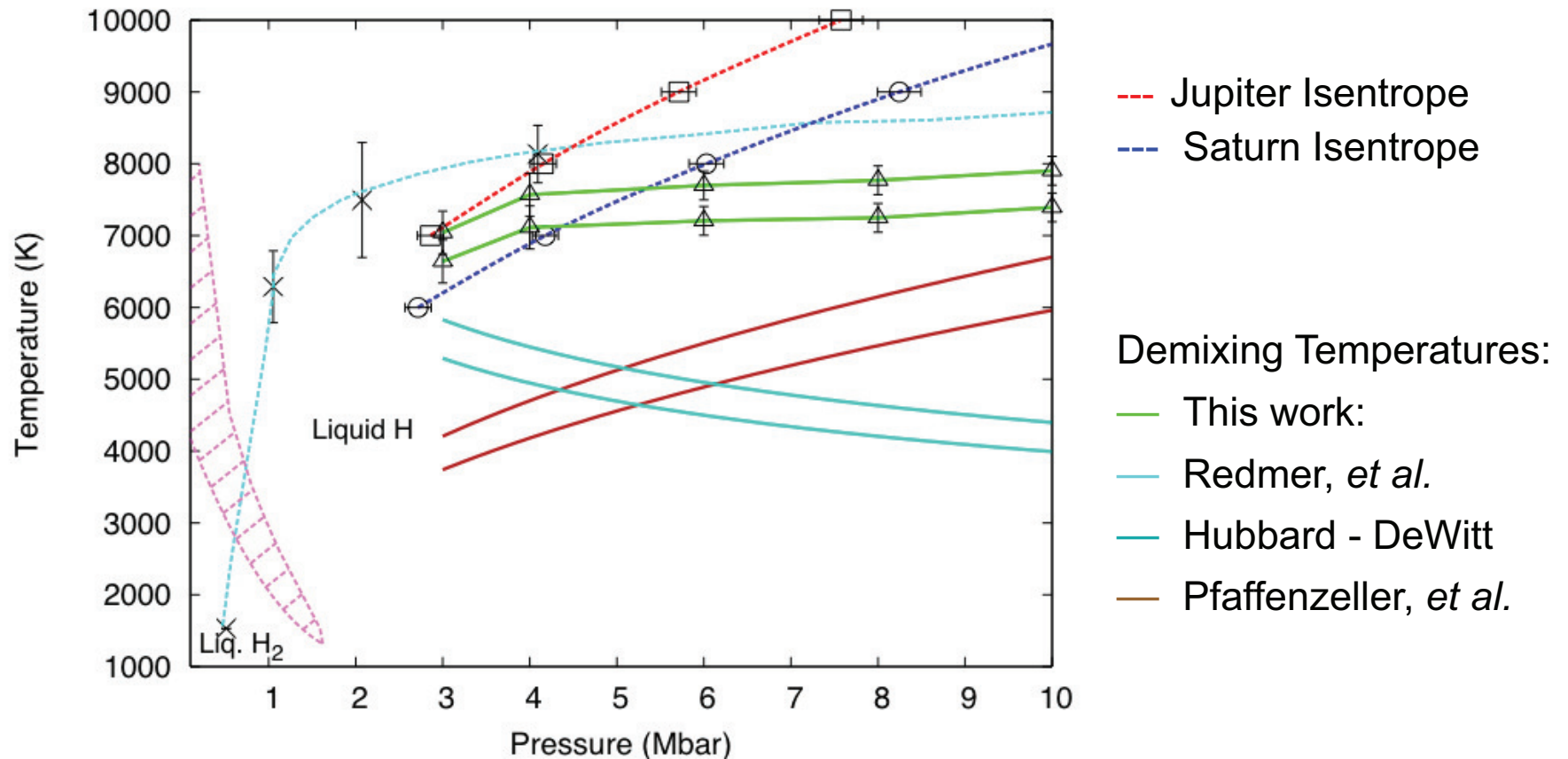


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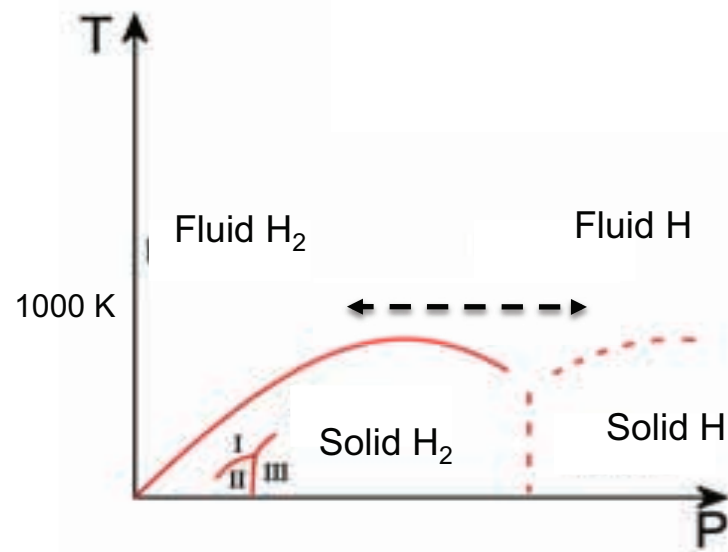
Ideal mixing works

Phase Diagram



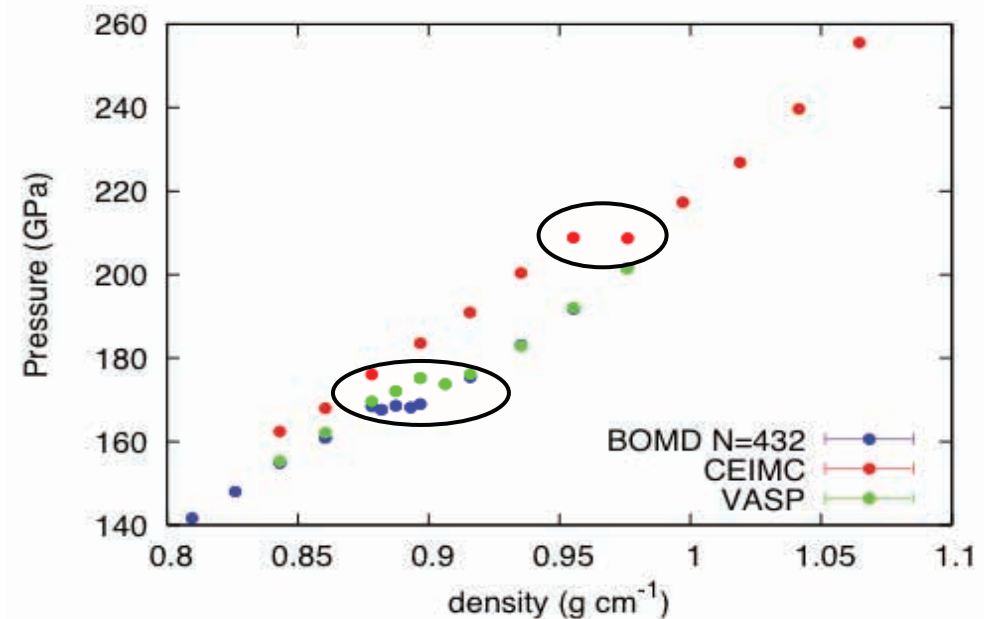
Morales M., *et al.*: PNAS 106:1324-1329 (2009)

Molecular Dissociation

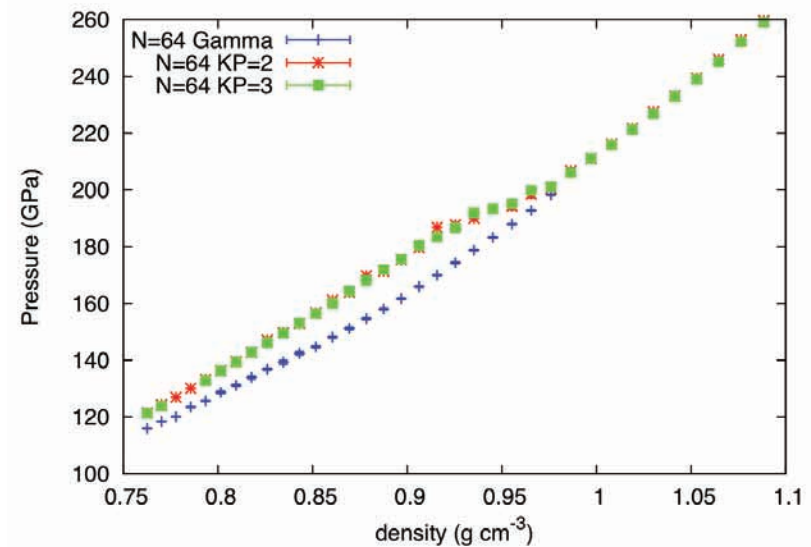
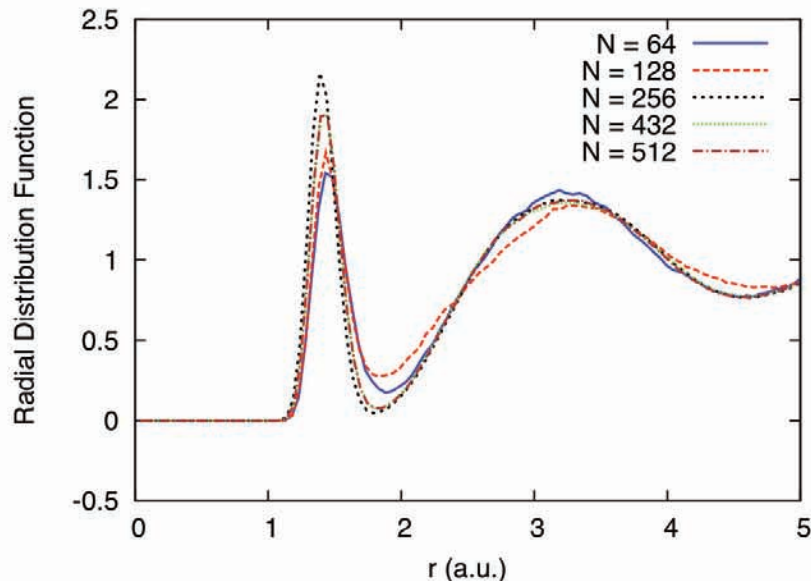


Liquid-liquid Phase Transition

- Pressure plateau at low temperatures
 - Clear signature of 1st order transition
 - Both CEIMC and BOMD simulations show signature
- Similar behavior observed at 2000 K, 1500 K and all the way down to 600 K.

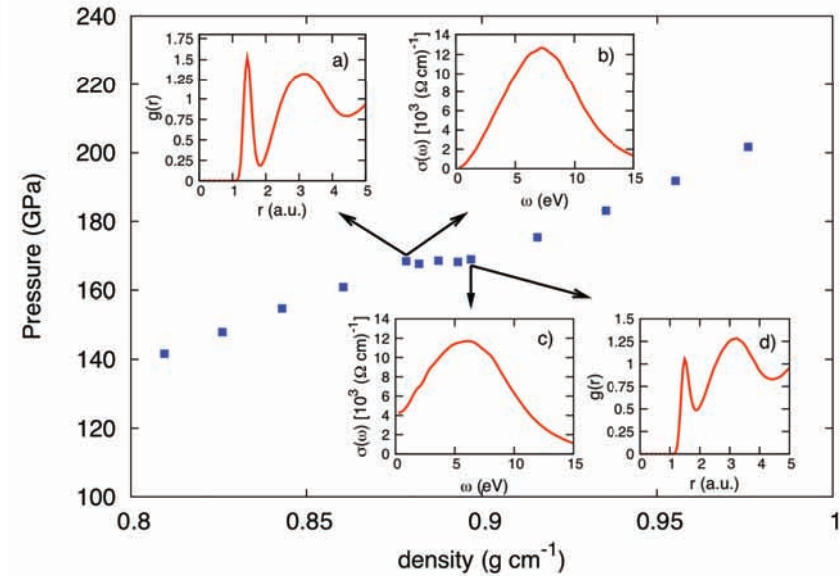
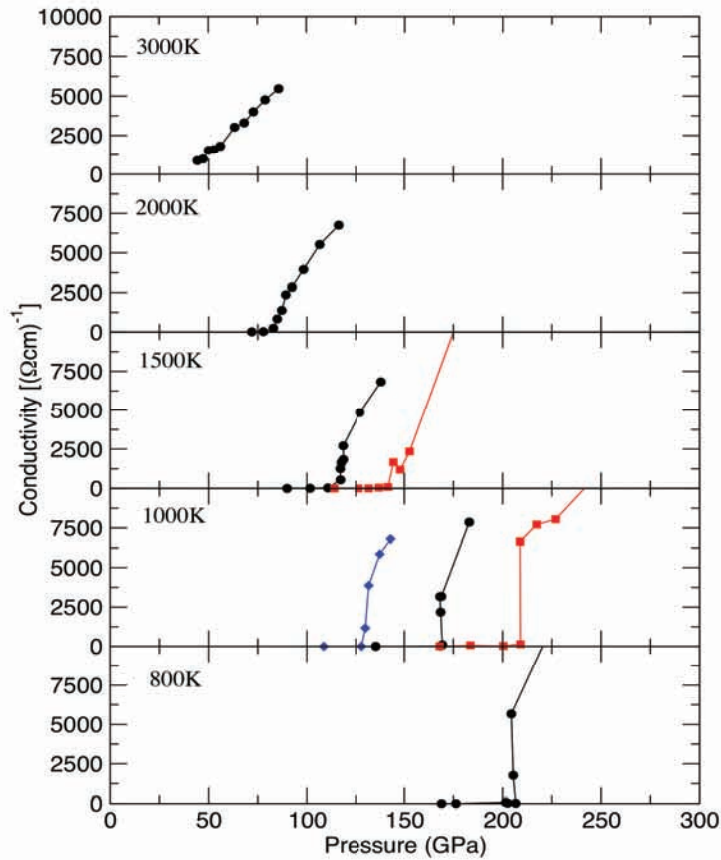


LLT - Size Effects



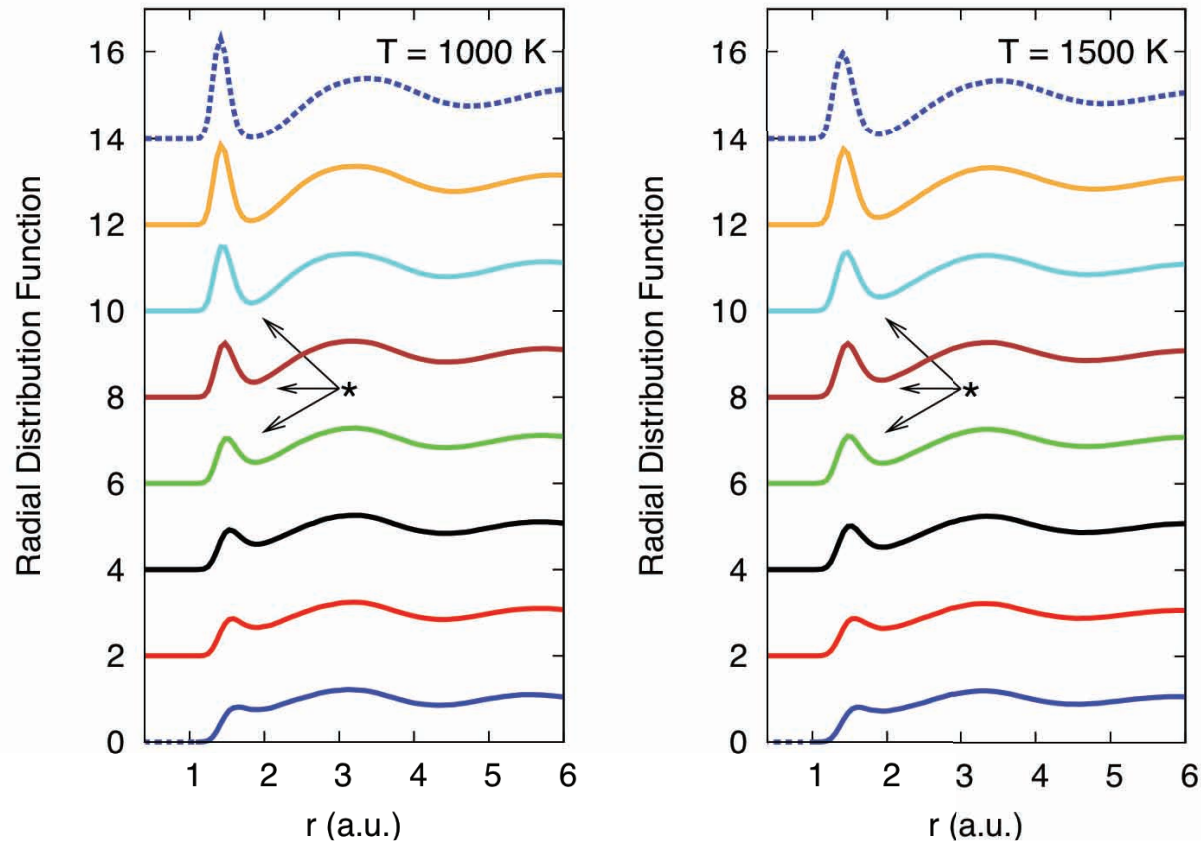
- Electronic size effects are very important in bonding
 - 64 atoms at Γ -point do not show transition at 1000 K
 - Transition appears with k-point integration

LLT - Electronic Conductivity



- Clear signs of sharp metallization across the transition
- Extrapolate discontinuity to find critical point

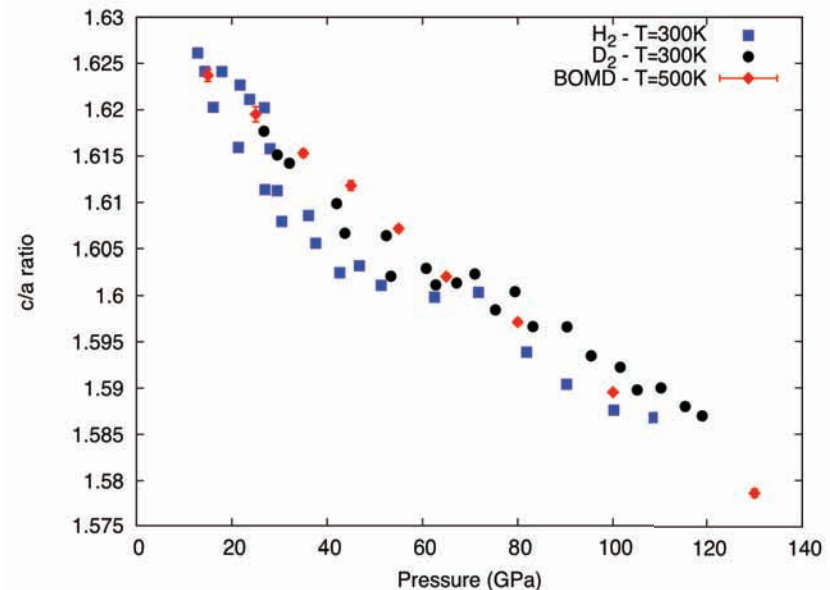
LLT – Molecular Structure



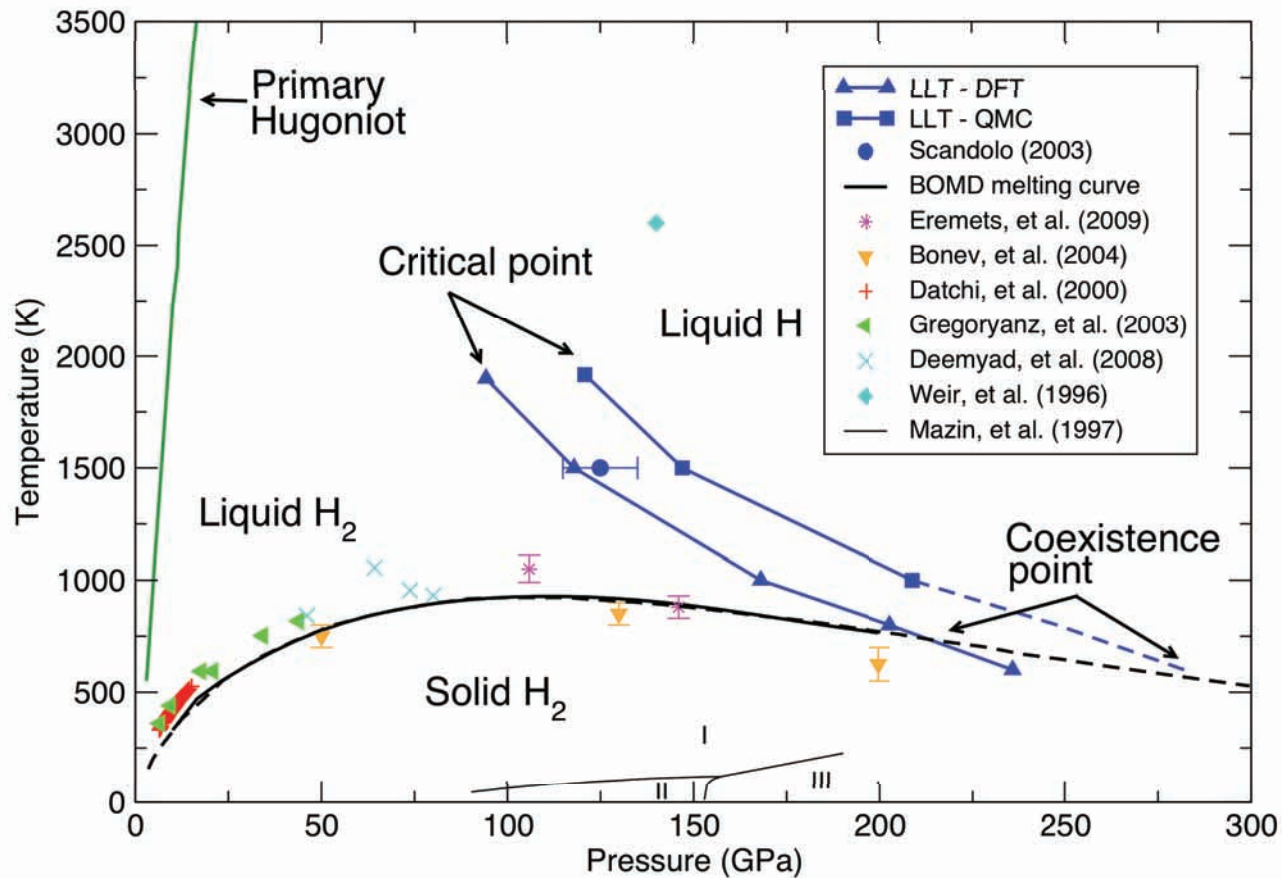
Strong short range correlations in atomic state

H₂ Melting Line

- Independent free energy calculation for liquid and solid phases
 - 10 GPa < P < 200 GPa
 - 500 K < T < 1500 K
- Solid Molecular phase
 - Assume Phase-I, hcp lattice without orientational order
 - NPT simulation to determine equilibrium lattice
 - Coupling constant integration using Einstein crystal method.



Revised Hydrogen Phase Diagram



Morales M. A., *et al.*, accepted for publication in PNAS.

Summary

- First principles EOS for metallic hydrogen from QMC
 - Good agreement with DFT-based simulations.
- Clear signature of helium immiscibility at high pressure.
 - Strong evidence for helium condensation in Saturn.
- Clear evidence for 1st order LLT in hydrogen
 - Critical point at $T \sim 2000$ K
 - Intersects melting line below $T \sim 800$ K, above 200 GPa.

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